

# Phyx 320

# Modern Physics

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March 15, 2021

Reading: 39.1 – 39.4

Homework #7 and Reading Reflection Thursday 11:59 pm

# Probability Density

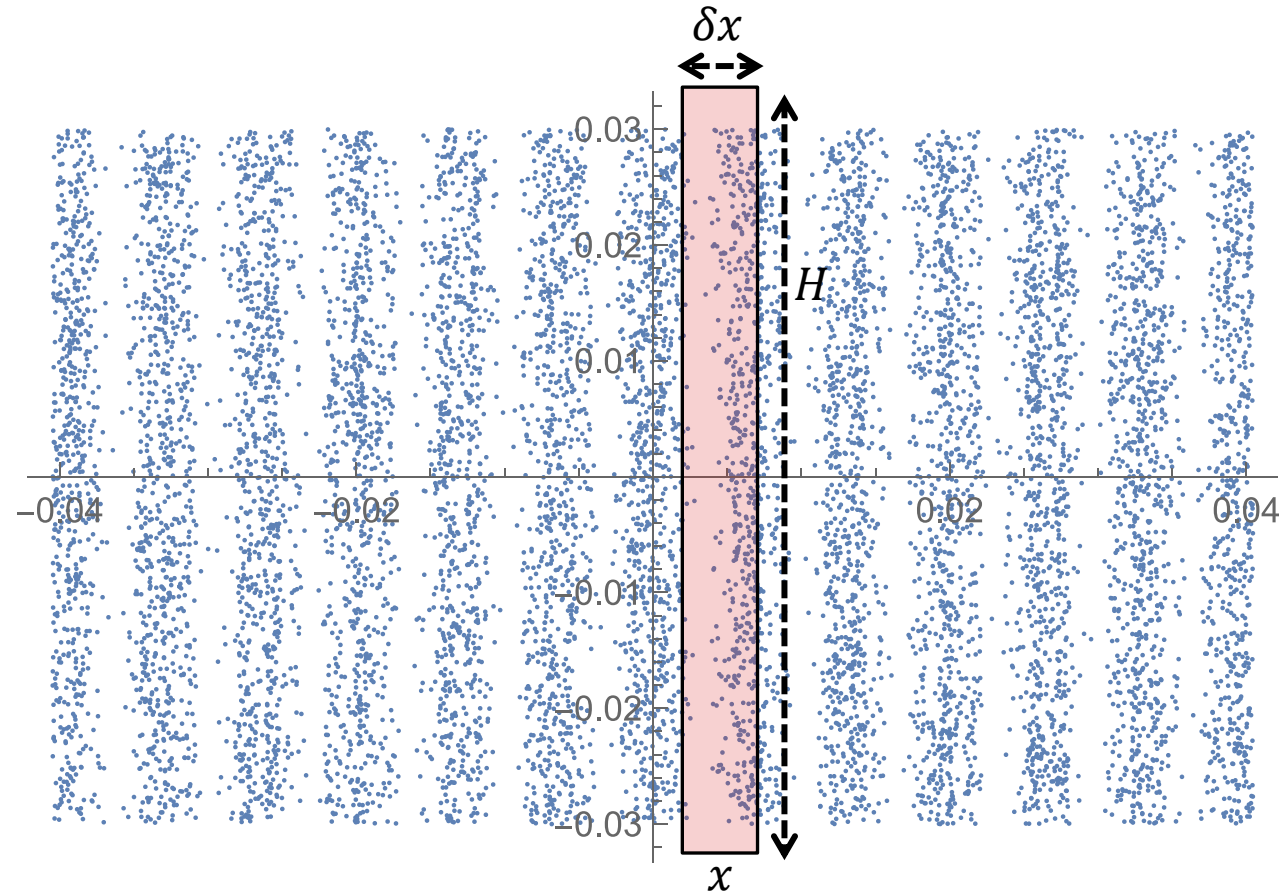
The probability density for photons is directly proportional to the square of the amplitude of the corresponding electromagnetic wave

$$P(x) \propto A(x)^2 \propto I(x)$$

Next lecture we will discuss the wavefunction which is generalization of this to every quantum particle

$$P(x) \propto |\psi(x)|^2$$

Wavefunction  
(complex number)



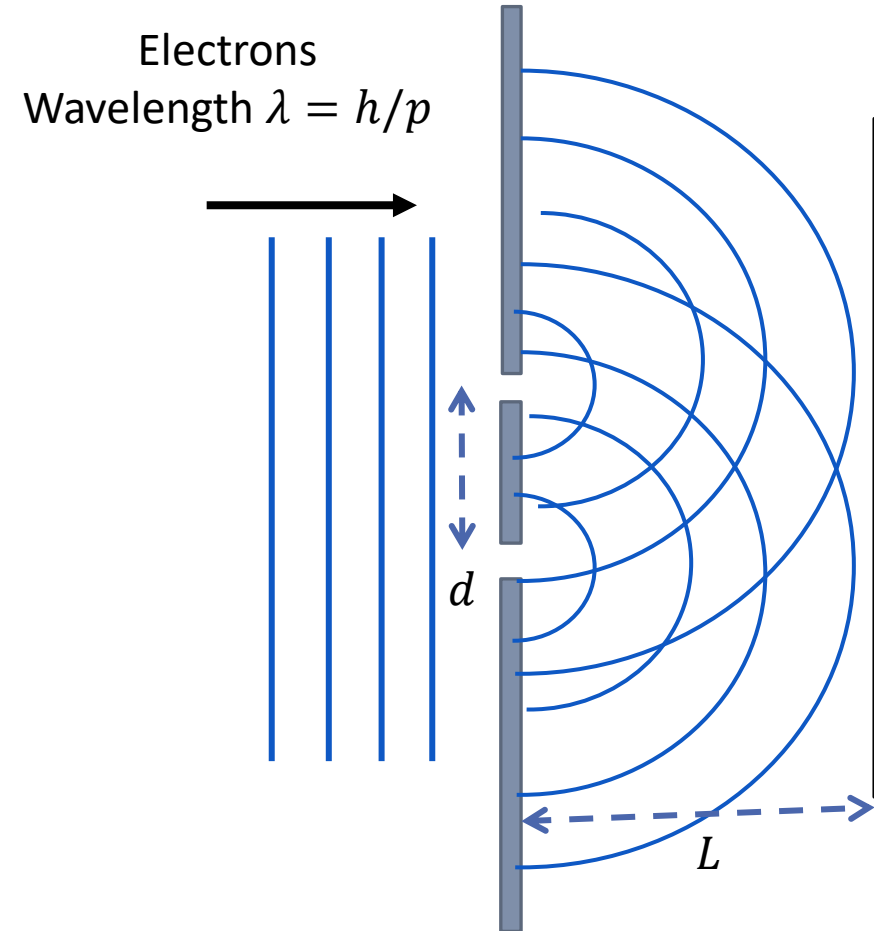
# Wave Function

To extend our discussion of the double-slit experiment to electrons, we assert that electrons are described by a wavefunction  $\psi(x)$

The probability density is then the wavefunction squared

$$P(x) = |\psi(x)|^2$$

Wavefunction  
(complex number)



# Wave Function

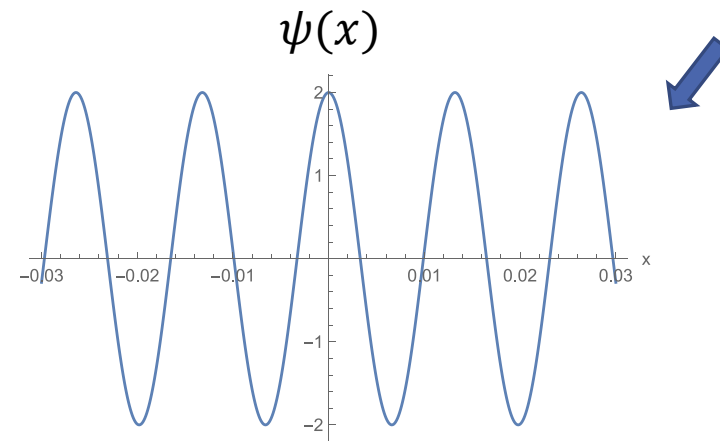
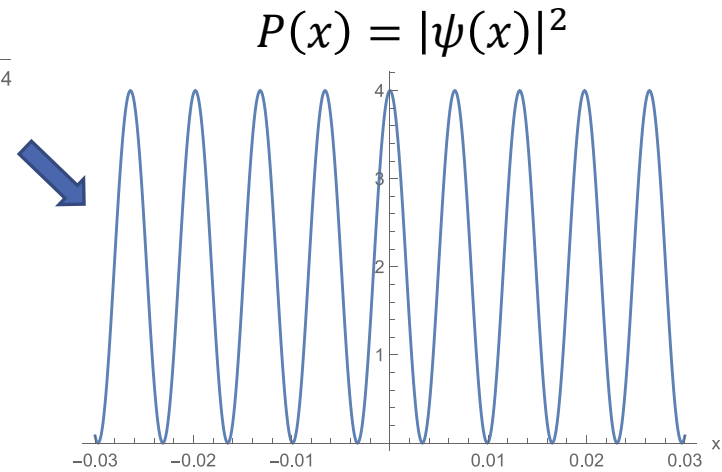
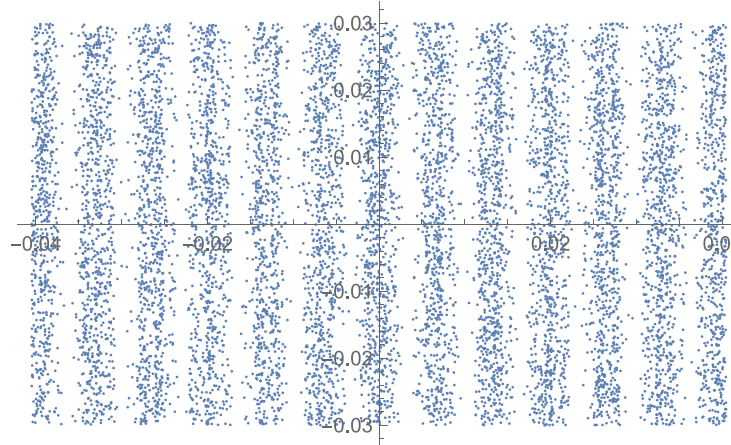
We can work backwards from the observed pattern of electrons to the probability density,  $P(x)$

Where  $P(x)$  has a maximum are locations more likely to find an electron

With many electrons going through the slits, more probability = more electrons

Wavefunction,  $\psi(x)$ , is square root of  $P(x)$

We can't observe  $\psi(x)$  directly so any function that yields the observed  $P(x)$  is a valid wavefunction ( $\psi(x) = -\psi(x)$ )



# Superposition

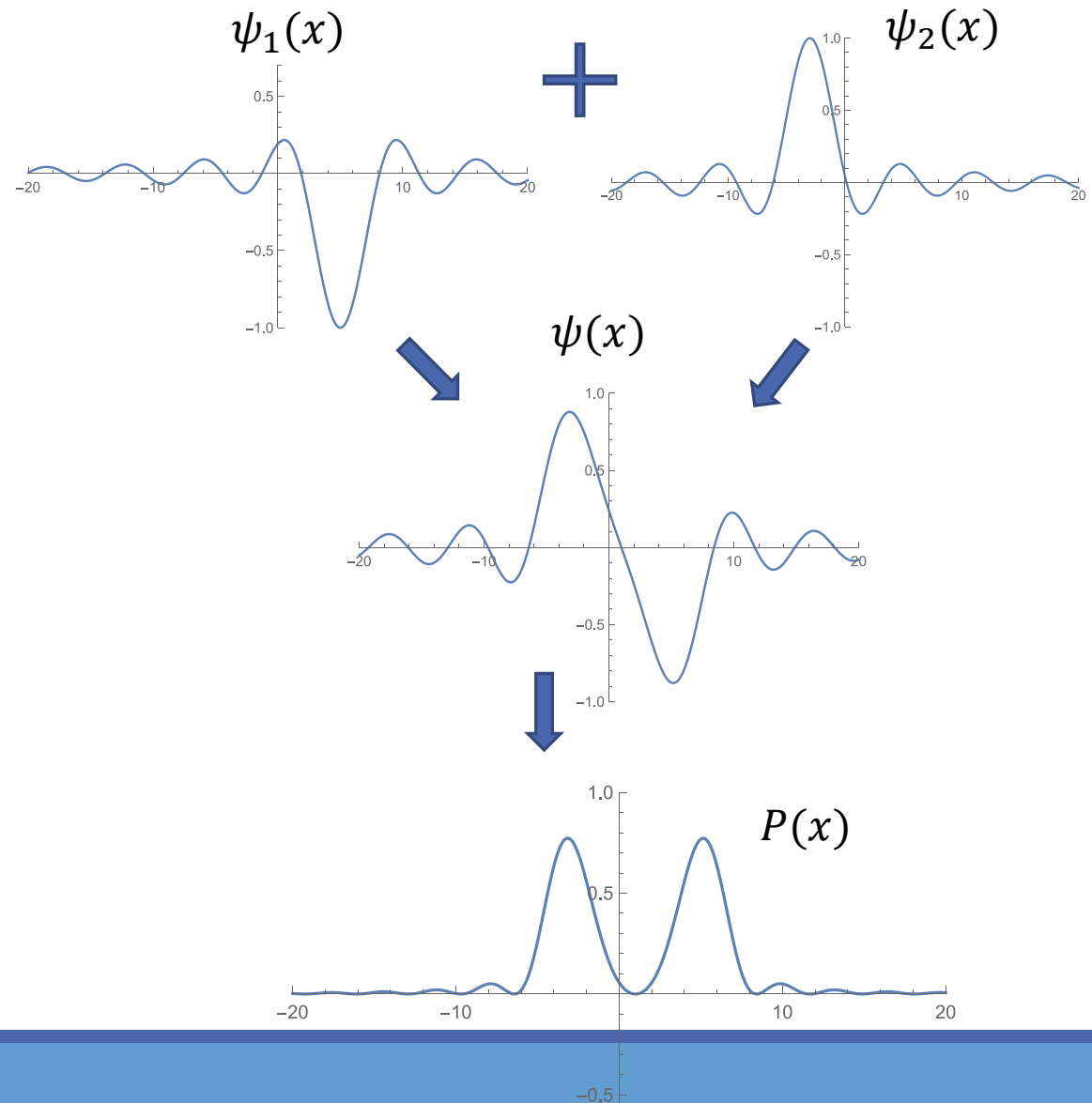
Wavefunctions add together just like waves

Quantum particles can be a superposition of multiple states

Let's say we have some system where the particle has equal likelihood of being in two states,  $\psi_1(x)$  or  $\psi_2(x)$

$$\psi(x) = \psi_1(x) + \psi_2(x)$$

For example, the electrons in the double-slit experiment are in a superposition of going through slit 1 and slit 2

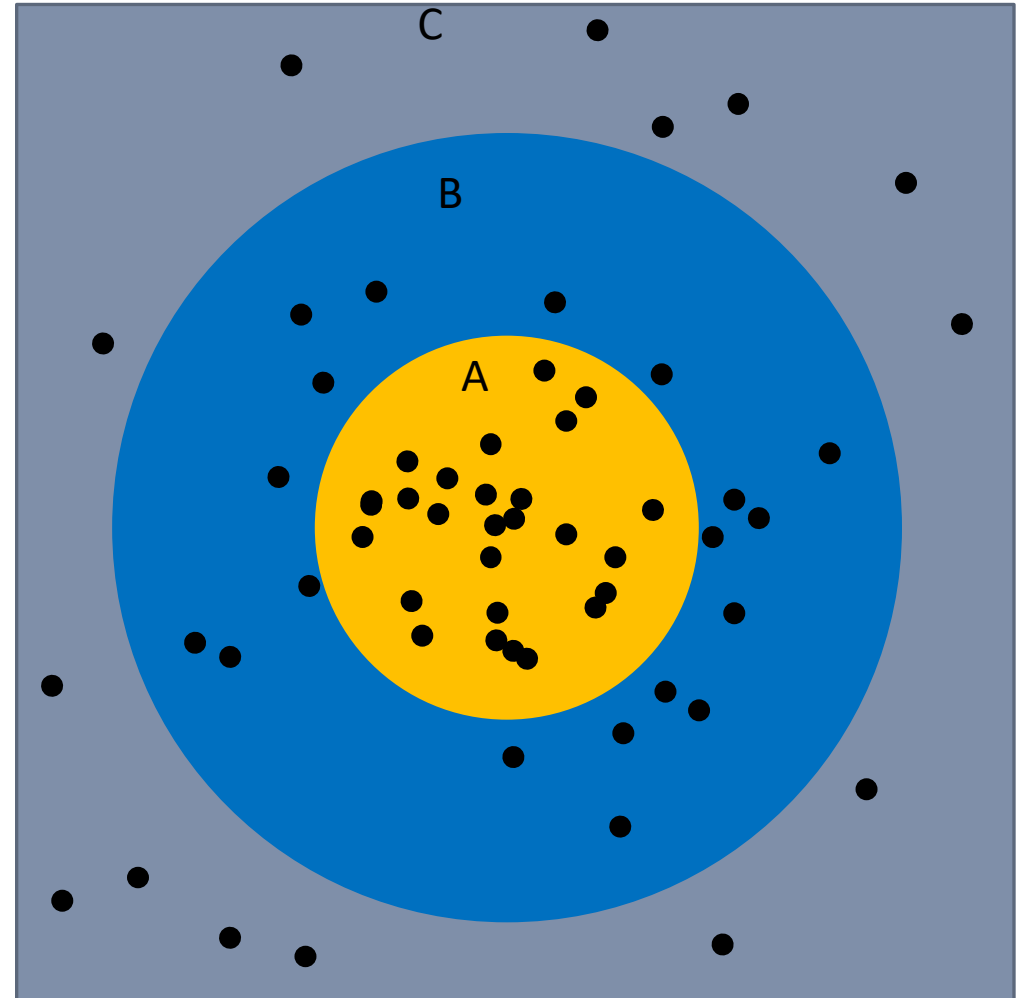


# Normalization

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In order to make logical sense, we have to force the probability of the particle landing somewhere to be 100%

Going back to dart board example

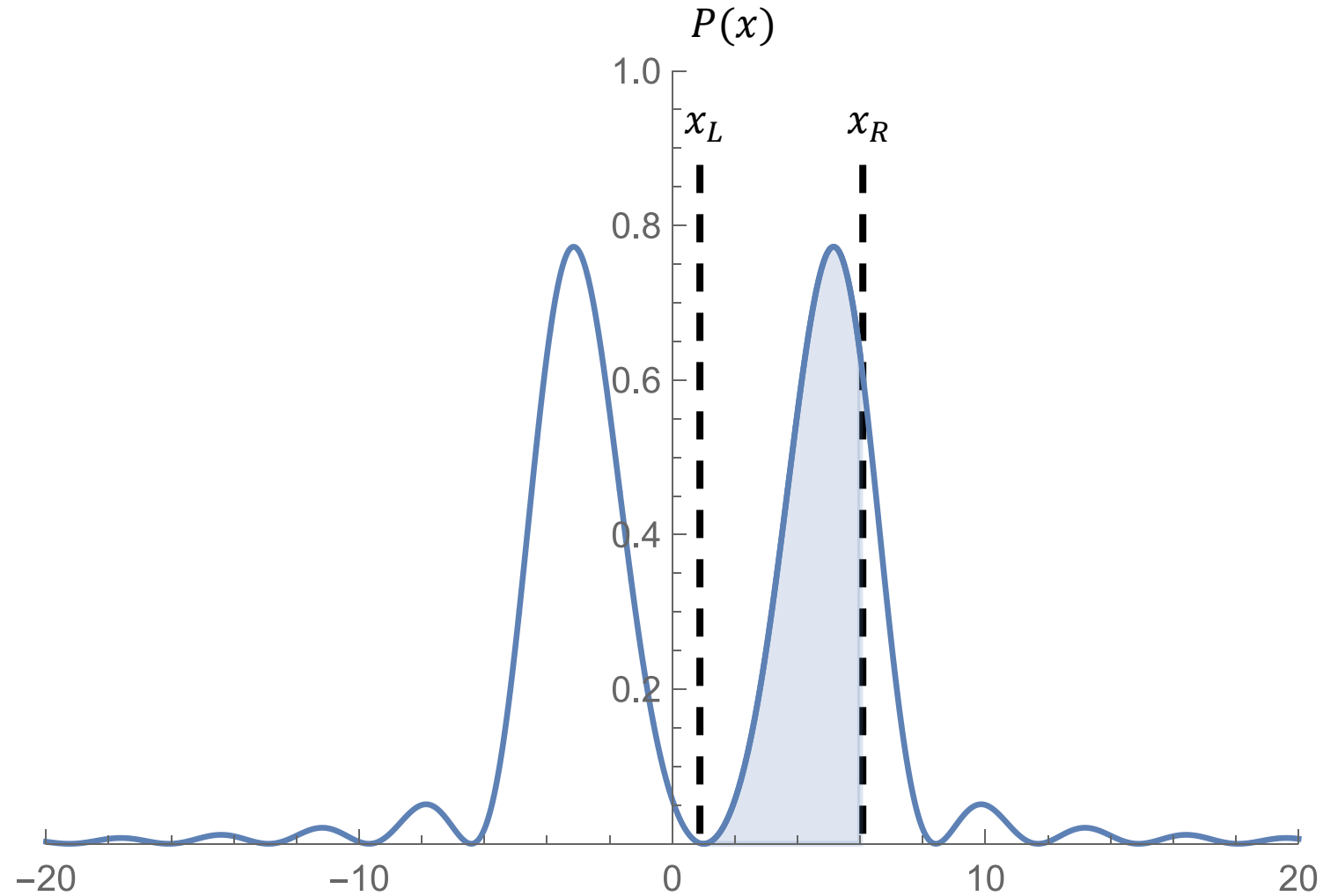


# Normalization

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Wavefunctions are continuous so we must integrate

Probability of finding particle in range  $(x_L, x_R)$

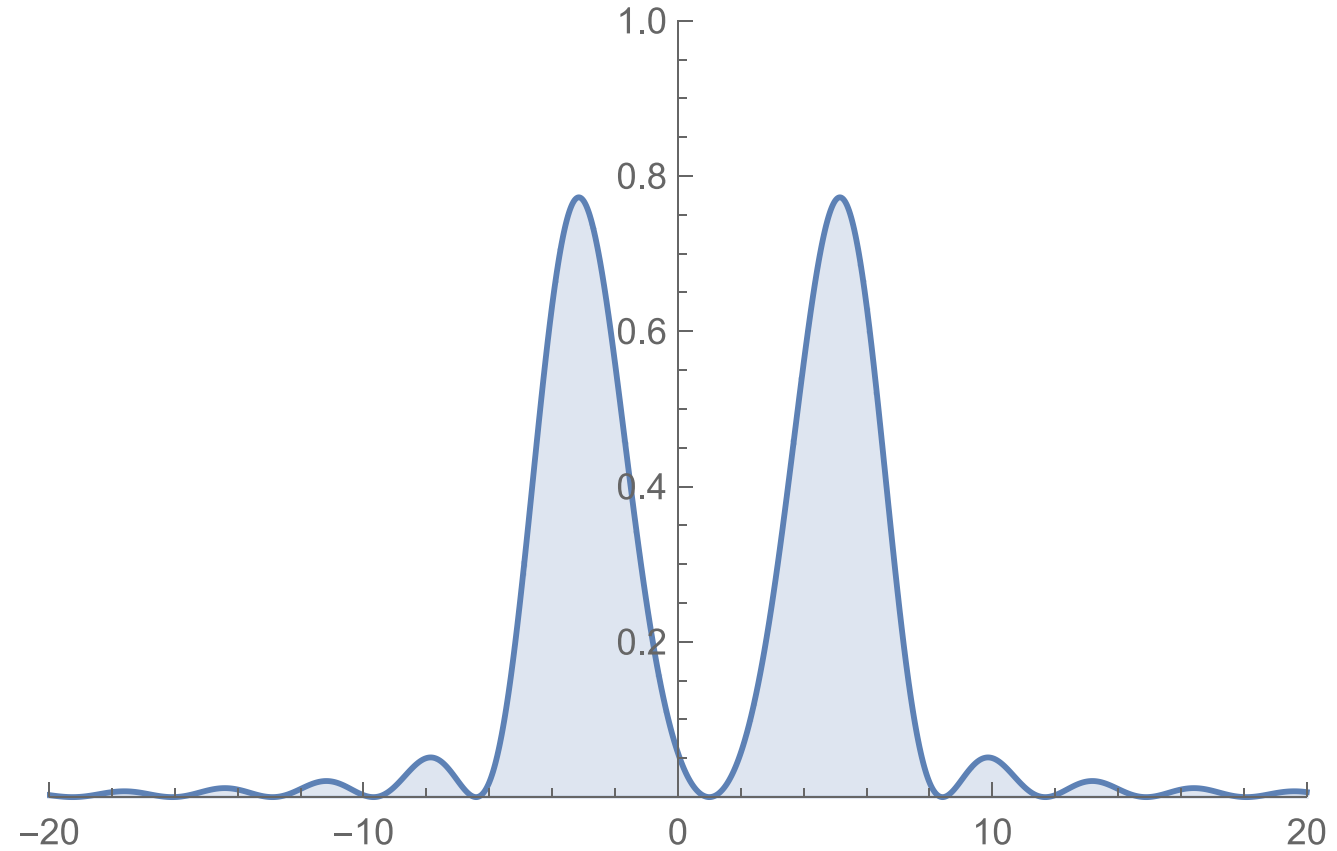


# Normalization

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If we integrate  $P(x)$  over all allowed values of  $x$ :  $(-\infty, \infty)$ , we must get 1

This is the same as saying the particle must be somewhere

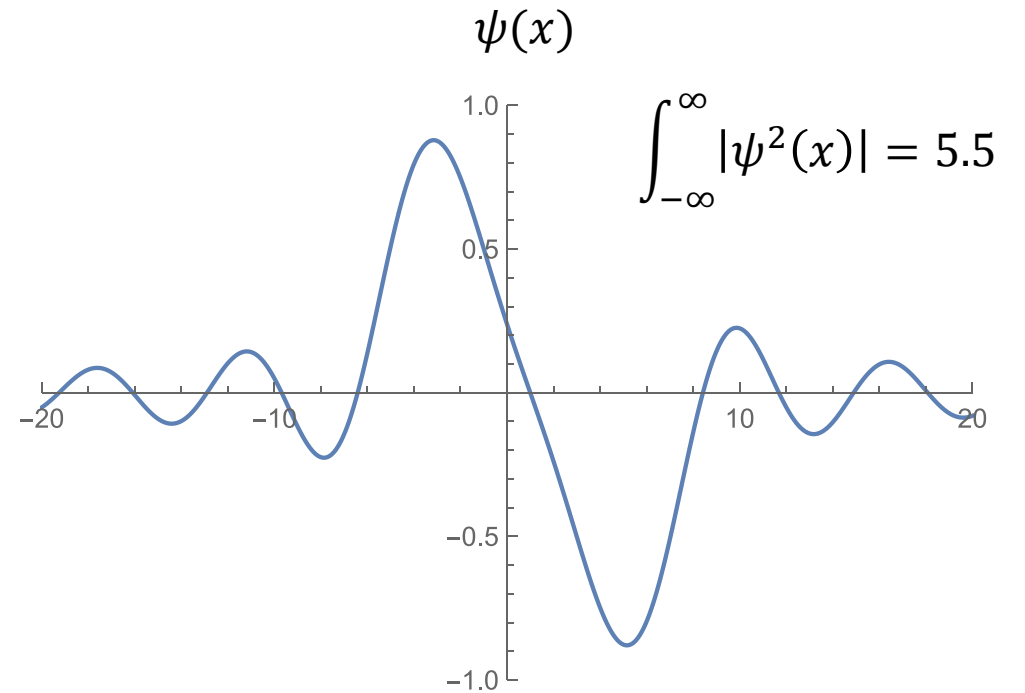




# Normalization

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What if we have a wavefunction that isn't normalized?



# Wavefunctions

Quantum particles (electrons, photons, protons, etc.) are described by wavefunctions  $\psi(x)$

Wavefunctions follow superposition so can be added  $\psi(x) = \psi_1(x) + \psi_2(x)$

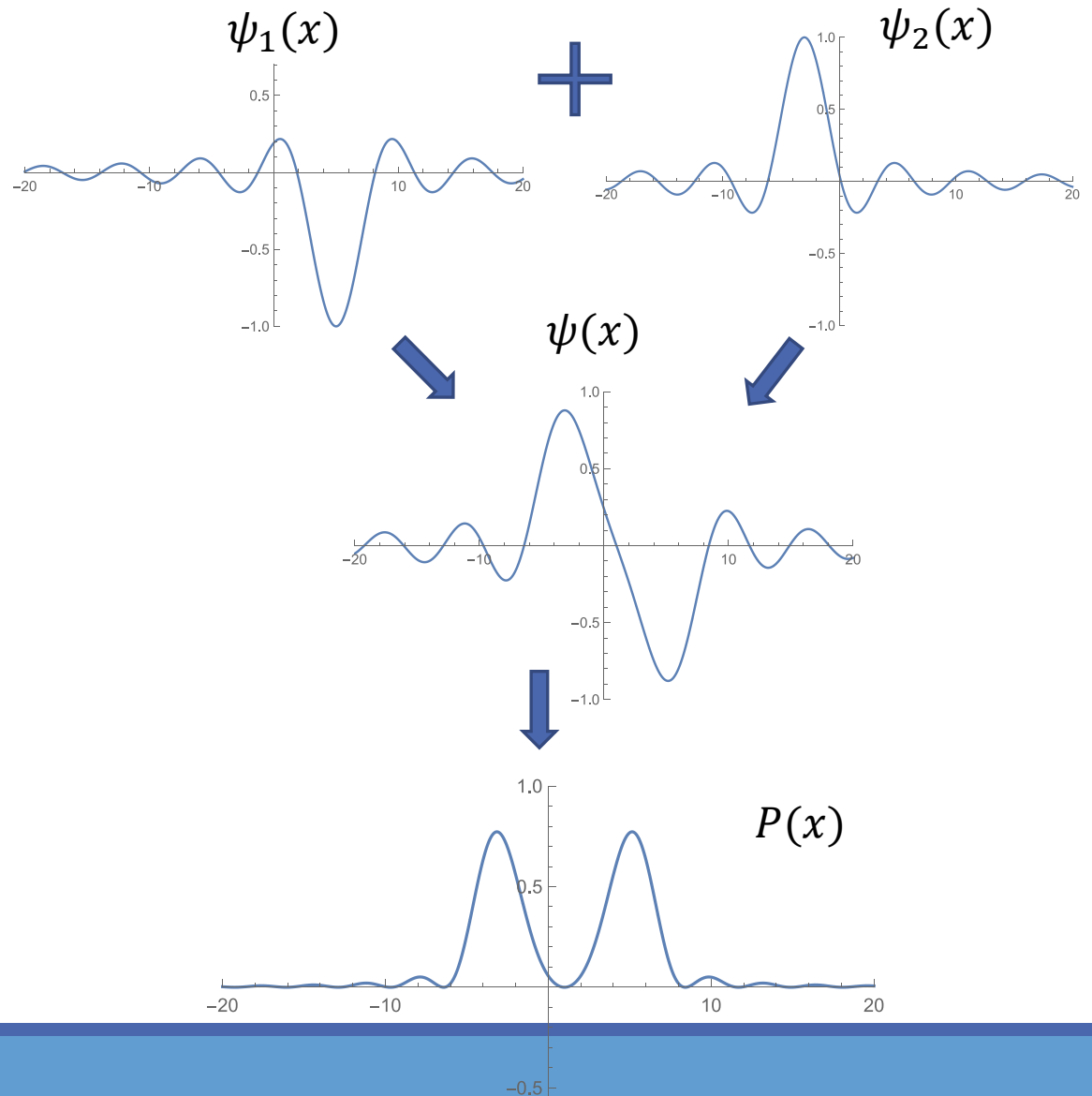
This produces interference effects

The probability density is defined to be

$$P(x) = |\psi(x)|^2$$

Wavefunctions must be normalized

$$\int_{-\infty}^{\infty} |\psi^2(x)| = 1$$



# Homework Questions

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